Optical Restoration of Lead Fluoride Crystals

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Lead Fluoride (PbF₂)

Purpose: Cherenkov Radiator

Cherenkov Radiation

- Photonic shock wave
- Particle velocity > c/n
- Wake of charged particle creates photons
- Little destructive interference → light cone



Advanced Test Reactor at INL

Cherenkov Radiation

Cherenkov cone is shown

 $\cos(\theta) = v_{\text{light}} / v_{\text{particle}}$

 $v_{light} = c/n$

n is index of refraction



n = 1 for vacuum; n = 1.00029 for air C = 2.99 x 10^8 m/s

Cherenkov Radiators

- In particle physics Cherenkov radiators afford means of distinguishing among particles (such as e⁻ vs proton). Particles may be identified by the angle of emitted Cherenkov light for a known index of refraction.
- In an investigation of scintillators in the late 80's, lead fluoride was found instead to be a Cherenkov radiator
- 132 blocks of PbF₂ will be used in the Deeply Virtual Compton Scattering experiments at Jefferson Lab (E07-007, E08-025, E12-06-114)

Lead Fluoride (PbF₂)

- Cherenkov Radiator
- Compact
- High Transparency
- Radiation Hard
- Low Hygroscopicity
- Optically restored after radiation exposure (UV/optical bleaching)



Cherenkov Radiators

Material	OPAL EM Calorimeter material					
	Pb	F5	SF5	SF57	Heavy	PbF2
PbO content (%)	1.77	-40	51	75	85	÷.
Density (g/cm³)	11.3	3.47	4.07	5.57	6.2	7.8
Index of refraction		1.6	1.6	1.85	55775	1.82
UV absorption edge (nm)	0.5	370	380	380	450	300
Radiation length (cm)	0.5	2	1.6	1.5	1.4	0.93
Interaction length (g/cm2)	193	104	108	120	120	156
Interaction length (cm)	18	29.9*	26.5*	21.4*	20**	20

* GEANT4 results ** Estimate

Zhao, Plannar Active Absorber Calorimeter, ACFA 2007

Shower Cascade

- Incoming electron interacts, releasing a photon via bremsstrahlung
- Photon pair produces $\gamma \rightarrow e^-e^+$
- Moving charged pair interacts again, producing more gammas
- Process continues, creating a shower effect
- Shower = big signal
- X₀ is radiation length, i.e. the mean distance over which electron beam intensity is reduced by a factor of e⁻¹



[Photo Reference] Gregoire, DVCS experiment, Ohio University 2007

Moliére Radius

- The Moliére Radius (R_M) is a measure of the size of an electromagnetic shower in some material
- Smaller Moliére radius = better shower containment
- For SF-5 and SF-6 lead glasses R_M = 3.7 cm & 2.7 cm, respectively¹
- The effective Moliére radius of lead fluoride² is 1.8 cm

[1] P. Kozma, R. Bajgar, P. Kozma Jr., Nucl. Instr. & Meth. A 484 (2002) 149
[2] P. Achenbach, I. Altarev, K. Grimm, T. Hammel, D. von Harrach, J. Hoffmann, H. Hofmann, E. –M. Kabuβ, S. Köbis, A. Lopes Ginja, F. E. Maas, E. Schilling, H. Ströher, Nucl. Instr. & Meth. A 416 (1998) 357

Our Experiment

- Measure transmission in samples of lead fluoride scheduled for use in the DVCS experiments at JLab
- Four samples exposed to 7-35 kGy of radiation under controlled, reproducible conditions
- The purpose was to quantify efficiency loss in DVCS detector

1 Gy = 100 rad = 1 J/kg (10-20 Gy of whole body dose can be fatal to humans)

The Method

- Dose samples with controlled source
- Measure transmission
- Optically bleach samples with blue light
- Measure transmission
- Repeat irradiation/optical restoration cycle
- Compare cycles

The Apparatus

- The source of exposure was the 25 MeV pulsed electron LINAC in the main hall of the Idaho Accelerator Center
- This facility offers adjustable pulse widths (80 ns 2 us) and repetition rates (1 Hz 1000 Hz) → predictable source
- Exposure may be monitored real-time with pick-up coil at the exit window
- Previous experiments^{1,2} used isotropic sources of gamma radiation





Calibration

- We wanted an accurate determination of dosage for all exposures from current monitor
- We tested the response of the current monitor with respect to a Faraday cup
- Check the proportionality of measurements of beam current to beam charge.



Coil Calibration



Faraday Cup saturates around 400 ns pulse width ~ 4.5×10^{12} electrons (Q = ldt)

Our Experiment



Dose Calculation

- Blocks were placed a distance from the exit window where it was known that the beam divergence (angular spread) would span only the face of a crystal (1 inch by 1 inch)
- Assuming that the total energy in the beam is absorbed in the block, dose may be calculated from beam pulses tracked by the pick-up coil
- Particles Per Second x Beam Energy x Time of Exposure = Total Energy Deposited

Example Calculation

- Beam intensity I depositing power P depends on energy E & electron charge e → P = E/e x I
- Dose = (P x Time of Exposure)/Mass
- P in watts, E in MeV and I in microamp
- For a window of 100 ns, repetition rate of 10 Hz, energy of 20 MeV and peak beam intensity of 50 mA we have

 $I = ((50 \text{ mA}) \times (100 \text{ ns}))/(10 \text{ Hz}) = 0.05 \text{ uA}$ P = E/e x I = 20 MV x 0.05 uA = 1 Watt For 1 Hour exposure of 1 kg mass, Dose = (1 W x 3600 s)/(1 kg) = 3.6 kGy

Color Center

 Lattice defects are introduced by exposure to high-energy photons (X, γ radiation) Negative ion vacancies attract electrons Since these electrons are bound to a positively-charged center, there is a spectrum of energy levels, as for atoms Excitations between energy levels result in optical absorption/emission spectra

Color Center

- Trapped electrons give rise to a discolored crystal
- This is qualitative evidence for the presence of color centers, absorbent portions of the lattice that hinder detection of the "good stuff" (Cherenkov light)
- This photo was taken just after exposure to radiation.



Transmission

- Transmission was measured along the length of each sample before and after exposure to violet/blue light
 Significant gains in transmission were
 - observed after 19 24 hours of optical curing



Before Optical Curing

- Data for the sample exhibiting the best response to optical bleaching is shown
- Transmission coefficient

 $T = I_{transmitted} / I_{incident}$

- Measurements were taken at 1 cm along the length of the sample – this region shows the biggest loss in transmission
- Color centers lower transmission efficiency significantly in the violet/blue region



After Optical Bleaching

- Additional spectra are transmission measurements after 19 hours of optical bleaching
- There are significant gains in violet/blue region
- Transmission is gained throughout entire spectral region



Why do we need blue?



Quantum Efficiency of Hamamatsu photomultiplier tube – peaks in blue

Optical Bleaching

- Out of place electrons in the lattice are endowed with enough energy from an outside light source to be repositioned
- In the barium fluoride experiment³, the most effective wavelengths (frequencies) were found to be 400-450 nm (violet/blue)
- Transmission measurements suggest that optical bleaching snaps the crystal lattice back into place, thus restoring its optical properties

[3] D. Ma, R. Zhu, On optical bleaching of barium fluoride crystals, Nucl. Meth. & Instr. A 332 (1993) 113

A Work In Progress

- Optical bleaching effective in PbF₂ for doses of up to 20 kGy
- Good for DVCS calorimeter
- Currently extracting n for comparison w/Fermilab⁴
- Spike in transmission at 410 nm (3.02 eV), 550 (2.25 eV) nm & 610 nm (2.03 eV)
- Objective prescriptive method for optical bleaching for DVCS calorimeter

[4] D.F. Anderson at al., Lead Fluoride: an Ultra-compact Cherenkov Radiator for EM Calorimetry, Nucl. Instr. & Meth. A 290 (1990) 385

Thank You

Extra Slides

(Questions?)

Index of Refraction

- Index of refraction as a function of wavelength, extracted from transmission data
- Agrees with measurements from Fermilab⁴ 20 years ago in the range of 400-700 nm



Energy Deposition

- Transmission data for same block at 16 cm along the length of the sample
- No significant difference in transmission is clear
- Color centers are formed closest to beam interface (the majority of the beam energy is deposited in the first few centimeters)



Contaminated Sample?

- Transmission data for block 16 at 16 cm is shown
- This sample shows anomalous behavior
- Spikes in transmission are suspected to be due to contaminants in this crystal

